

A New Class of Sweet Potato Phytoalexins

Josef A. Schneider and Koji Nakanishi*

SUNBOR (Suntory Institute for Bioorganic Research), Wakayamadai, Shimamoto-cho, Mishima-gun, Osaka 618, Japan

Two new compounds, 7-hydroxycostal (**2**) and 7-hydroxycostol (**3**), were isolated from infected sweet potatoes as members of a new class of sweet potato phytoalexins.

When the sweet potato *Ipomea batatas* is infected with the pathogenic black rot fungus *Ceratocystis fimbriata*, ipomeamarone (**1**) rapidly accumulates in the infected tissue. Isolated forty years ago by Hiura,¹ and characterized by Kubota and Matsuura,² the biochemistry of (**1**) has been extensively studied.^{3,4} In the accompanying communication the absolute configuration of (**1**) has been established as 1*R*,4*S*.⁵

In the course of our investigation of sweet potato phytoalexins, the infected tissue extract (CHCl₃) was found to contain a further class of defence compounds, namely sesquiterpenes composed of the eudesmane skeleton. Chromatography

of the extract (15 g) on silica gel afforded two new compounds, 7-hydroxycostal (**2**) (450 mg),[†] and 7-hydroxycostol (**3**) (50

[†] Physical data for (**2**) (C₁₅H₂₂O₂): *R*_T (20% EtOAc-hexane, silica gel) 0.40; [α]_D²⁰ + 15.0° (c 2.0, CH₂Cl₂); λ_{max} 215 nm, ε 5400; mass spectrum (electron impact) *m/z* (rel. intensity): 234 (*M*⁺, 2%), 216 (100), 202 (85), 145 (25), 83 (52), and 55 (8); i.r. (neat): 3480, 2950, 1695, 1645, 1620, 1440, 1305, 1055, and 885 cm⁻¹; ¹H n.m.r. (360 MHz, CDCl₃): δ 9.57 (s, 13-H), 6.53 (s, 12-H), 6.01 (s, 12-H), 4.68 (br. s, 14-H), 4.32 (br. s, 14-H), 2.95 (br. s, OH), 2.36 (br. d, *J* 12.5 Hz, 5-H), 2.29 (dt, *J*₁ 14.0, *J*₂ 3.0 Hz, 3β-H), 2.02 (br. dt, *J*₁ 13.5, *J*₂ 7.0 Hz, 3α-H), 1.92 (dt, *J*₁ 14.0, *J*₂ 4.5 Hz, 9α-H), 1.78–1.24 (m, 9 protons), and 0.73 (s, 10-Me).

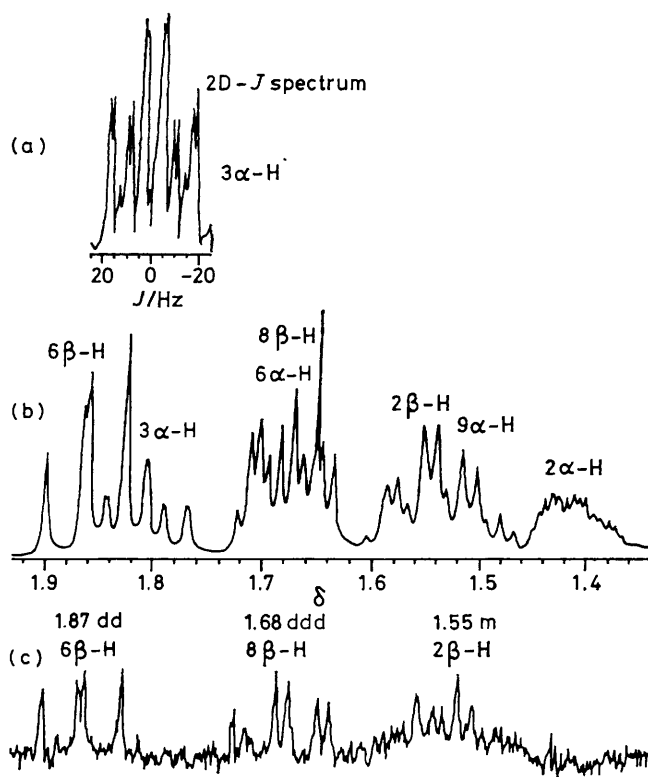
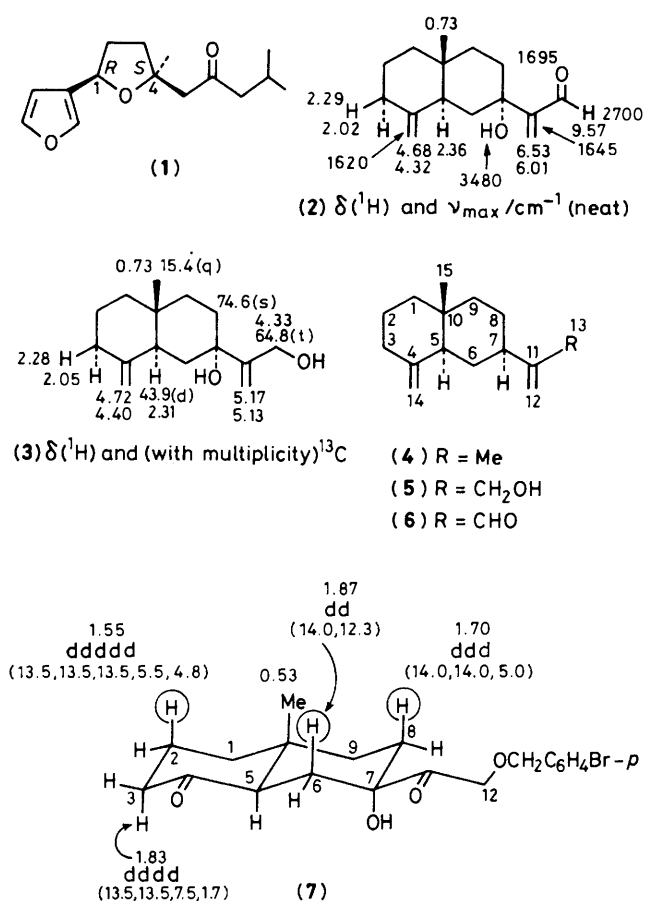


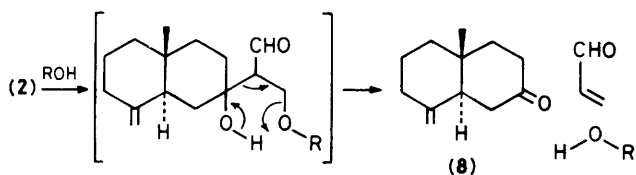
Figure 1. 360 MHz N.m.r. spectrum of (7) (^1H ; C_6D_6): (a) 2D J resolution of $3\alpha-5\alpha$ coupling; (b) normal spectrum; (c) difference in n.O.e. spectrum (irradiating 10-Me).

^1H N.m.r. resonances for (7) and coupling constants (J/Hz); see Figure 1.

mg). \ddagger These were accompanied by the known selinene derivatives (4), (5), and (6).⁶

The structure of (3) was determined by studying its dioxo-derivative (7) (i, $p\text{-BrC}_6\text{H}_4\text{CH}_2\text{Cl}$, pyridine; ii, $\text{O}_3\text{-Me}_2\text{S}$), which was more amenable to n.m.r. studies. Application of 2 dimensional J spectroscopy,⁷ together with double resonance experiments permitted the assignment of most protons and coupling constants (Table 1). In addition to the usual W-type interactions, a 1.7 Hz 1,3 diaxial coupling⁸ between $5\alpha\text{-H}$ and $3\alpha\text{-H}$ was resolved in the 2D J spectrum (Figure 1a). Irradiation of 10-Me and measurement of the nuclear Overhauser effect (n.O.e.) difference spectrum isolated the $2\beta\text{-}$, $6\beta\text{-}$, and $8\beta\text{-H}$ signals (Figure 1c) from other overlapping signals (Figure 1b).

The stereochemistry at C-7 was determined as $\alpha\text{-OH}$ on the basis of the downfield shift of $5\alpha\text{-H}$ in (2) and (3), δ 2.36 and 2.31, respectively; in compounds (5) and (6) where the $\alpha\text{-OH}$ group is replaced by hydrogen, $5\alpha\text{-H}$ resonates at δ



Scheme 1. Mechanism for formation of (8). R = Me or H (see text).

1.83 and 1.88, respectively. The isolation of $\beta\text{-selinene}$ (4), $[\alpha]_D + 45^\circ$ (c 5.0, CH_2Cl_2), lit.,⁹ $+ 49^\circ$, and $\beta\text{-costol}$ (5), $[\alpha]_D + 26.8^\circ$ (c 2.0, CH_2Cl_2), lit.,¹⁰ $+ 32.8^\circ$, from the same extract provided the basis for assigning the $5R,7S,10R$ -configuration in (2) and (3). \S

Incubation experiments in sweet potato culture medium containing endoconidia of *C. fimbriata*,¹⁰ showed (2) to be a more effective germination inhibitor than (+)-ipomeamarone (1). Interestingly, the 7-deoxy-aldehyde (6) was essentially devoid of activity. Clearly, this difference in antifungal properties must be attributable to hydroxylation at C-7. We reasoned that the presence of the 7-OH moiety may facilitate a retro-aldol process to give the ketone (8) and acrolein (Scheme 1). Indeed, after incubation for one week, a *ca.* 10% conversion of (2) to yield largely (8) was observed. \P We therefore propose

\ddagger Physical data for (3) ($\text{C}_{15}\text{H}_{24}\text{O}_2$): R_f (20% EtOAc-hexane, silica gel) 0.35; m.p. 123–124 $^\circ\text{C}$; $[\alpha]_D + 2.0^\circ$ (c 0.4, CH_2Cl_2); mass spectrum (field desorption) m/z (rel. intensity): 236 (M^+ , 40%) and 218 (100); i.r. (neat): 3450, 2950, 1640, 1040, 890, and 870 cm^{-1} ; ^1H n.m.r. (360 MHz, CDCl_3): δ 5.32 (br. s, 12-H), 5.17 (br. s, 12-H), 4.72 (br. s, 14-H), 4.40 (br. s, 14-H), 4.33 (br. s, 13-H), 2.31 (br. d, J 13.5 Hz, $5\alpha\text{-H}$), 2.28 (br. d, J 14.0 Hz, $3\beta\text{-H}$), 2.12 (br. s, OH), 2.05 (br. dt, J_1 13.5, J_2 7.0 Hz, $3\alpha\text{-H}$), 1.85–1.30 (m, 11 protons), and 0.73 (s, 10-Me); ^{13}C n.m.r. (25.0 MHz, CDCl_3): δ 154.8 (s), 150.5 (s), 110.9 (t), 105.3 (t), 74.6 (s), 64.8 (t), 43.9 (d), 41.5 (t), 36.9 (t), 36.1 (t), 35.6 (t, 2 carbons), 32.5 (t), 23.4 (t), and 15.4 (q) p.p.m.

\S That (2) and (3) differ only in oxidation state at C-13 was established by converting (2) into (3) by oxidation with $\text{CrO}_3 \cdot 2\text{pyridine}$.

\P Treatment of (2) with a catalytic amount of sodium in methanol gave a 90% yield of (8).

Table 1. ^1H Chemical shifts and coupling constants (J/Hz) for (7) (360 MHz, C_6D_6).

δ				J <i>ax.-eq.</i>		J <i>gem.</i>		J <i>eq.-eq.</i>		J <i>ax.-ax.</i>	
1α	1.19	6β	1.87	$1\alpha 2\alpha$	4.8	$1\alpha 1\beta$	13.5	$1\beta 2\alpha$	3.0	$1\alpha 2\beta$	13.5
1β	1.10	8α	1.25	$2\beta 1\beta$	4.8	$2\alpha 2\beta$	13.5	$2\alpha 3\beta$	2.7	$2\beta 3\alpha$	13.5
2α	1.42	8β	1.70	$2\beta 3\beta$	5.5	$3\alpha 3\beta$	13.5	$9\beta 8\alpha$	3.0	$5\alpha 6\beta$	12.3
2β	1.55	9α	1.56	$3\alpha 2\alpha$	7.5	$6\alpha 6\beta$	14.0	long range		$8\beta 9\alpha$	14.0
3α	1.83	9β	0.99	$5\alpha 6\alpha$	3.5	$8\alpha 8\beta$	14.0	$1\beta 3\beta$	1.5	n.o.e. (irr. 10-Me)	
3β	2.17	10-Me	0.53	$8\beta 9\beta$	5.0	$9\alpha 9\beta$	14.0	$6\alpha 8\alpha$	0.5	2β	2.5%
5α	2.48	12-H	3.89	$9\alpha 8\alpha$	4.5			$3\alpha 5\alpha$	1.5	6β	2.7%
			7.13							8β	3.0%
6α	1.68	Ar-H	7.83								

that the release of acrolein accounts for the fungicidal action of (2). While (2) is present in approximately only one-sixth of the quantity of (1), it is obviously an important phytoalexin.

The sweet potato thus appears to represent a rare case of a plant capable of concurrently producing two skeletally different series of antifungal defence compounds.

We thank Professor I. Uritani and Dr. K. Oba of Nagoya University for providing us with infected sweet potato extract, and for helpful discussions. We also thank Dr. T. Iwashita of SUNBOR for 360 MHz n.m.r. measurements.

Received, 29th December 1982; Com. 1487

References

- 1 M. Hiura, *Rep. Gifu Agric. Coll.*, 1943, **50**, 1.
- 2 T. Kubota and T. Matsuura, *Chem. Ind. (London)*, 1956, 521; T. Kubota, *Tetrahedron*, 1958, **4**, 68.
- 3 I. Uritani, K. Oba, A. Takeuchi, K. Sato, H. Inoue, R. Ito, and I. Ito, 'Antinutrients and Natural Toxicants in Foods,' ed. R. L. Ory, Food and Nutrition Press, Westport, CT, p. 1.
- 4 L. T. Burka, L. J. Felice, and S. W. Jackson, *Phytochemistry*, 1981, **20**, 647, and references therein.
- 5 J. A. Schneider, K. Yoshihara, and K. Nakanishi, *J. Chem. Soc., Chem. Commun.*, preceding communication.
- 6 T. K. Devon and A. I. Scott, 'Handbook of Naturally Occurring Compounds, Vol. II, Academic Press, New York, 1972, pp. 137, 138.
- 7 A. A. Maudsley, A. Kumar, and R. R. Ernst, *J. Magn. Reson.*, 1977, **28**, 463; L. D. Halland and J. K. M. Sanders, *J. Am. Chem. Soc.*, 1980, **102**, 5703.
- 8 L. M. Jackman and S. Sternhell, 'Applications of Nuclear Magnetic Resonance Spectroscopy in Organic Chemistry,' Pergamon Press, Elmsford, NY, 1969, p. 338.
- 9 K. Yoshihara and Y. Hirose, *Bull. Chem. Soc. Jpn.*, 1978, **51**, 3395.
- 10 A. S. Dawdekar, G. R. Kelkar, and S. C. Bhattacharya, *Tetrahedron*, 1967, **23**, 1993.
- 11 M. Kojima and I. Uritani, *Phys. Plant Pathol.*, 1976, **8**, 97.